

Gibbs Adsorption in Binary Liquid Mixtures: Concentration Variations Beyond Surface Monolayer in BiSn alloy

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Abstract

- <u>Lead-Free Solders:</u>
- -Technological importance
- -Basic understanding of surface structure and thermodynamics of two-component liquid alloy
- Experiment & Theory:
- -Resonant X-ray measurements
- -Probing surface layering & composition
- -Comparison of results with Gibbs theorem and Defay-Prigogine model



Fresnel normalized x-ray reflectivities measured at four different energies, the lines show the best fit obtained using a three layer surface model.

Inset ("Δ" energies used for measurements):

- Bi L3 absorption edge @ E = 13.418 keV
- Right scale shows electron density contrast

MOTIVATION

Practical:

- Ban of Pb-based solders in industrial production due to health risks
- Need to replace PbSn
- solders by Pb-free alternatives
- Appealing candidate BiSn:
- **★**Low melting temperature
- **★**No health hazards

★ Inexpensive

Fundamental:

- Lack of understanding of surface thermodynamics of binary alloys: studied so far -BiIn, GaIn, GaBi, AuSi, AuGe, HgAu - behave differently!
- How do the properties of the two components define the surface of the binary alloy?



Summary

- •Studies of Pb-free solders such as BiSn are of great importance for industry and fundamental science
- •BiSn shows well-defined layering and composition variations at the topmost three layers
- •BiSn is the first system that shows atomic scale demixing proven by X-ray reflectivity measurements
- •Surface segreation and surface demixing can be explained within the Defay-Prigogine approach

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-Gibbs absorption theorem (assumes ideal solution) predicts a monolayer excess of Bi to reduce surface tension and surface energy

($\gamma(Bi) = 378 \text{ mN/m}, \gamma(Sn) = 560 \text{mN/m}^*$)

-extended Defay-Prigogine model (assumes regular solution) provides corrections for the oversimplified Gibbs absorption theorem

-BiSn data shows good agreement with Defay-Prigogine model and explains surface demixing within the first three layers

* surface tension @ melting temperature